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principles of carbon dioxide equilibria, presents casestudies demonstrating the ways in which carbon dioxide equilibria are used in physiology and oceanography, and explores

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Knowledge of thermodynamics, reaction quotient and Subject : CHEMISTRY Topic : IONIC EQUILIBRIUM Page 4 of 24 IONIC EQUILIBRIUM (b) A weak base and a strong base $[H^+]$ is entirely due to dissociation of strong base Neglect the contribution of weak acid/base usually.

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When a sparingly soluble ionic solid is dissolved in water it undergoes ionization and the ions forming from solid phase pass into the solution till the solution becomes saturated and a dynamic equilibrium is established between the ions present in the saturated solution and the ions present in the solid phase at a constant temperature.